

Kinetic model for the finite-time thermodynamics of small heat engines

Angelo VULPIANI

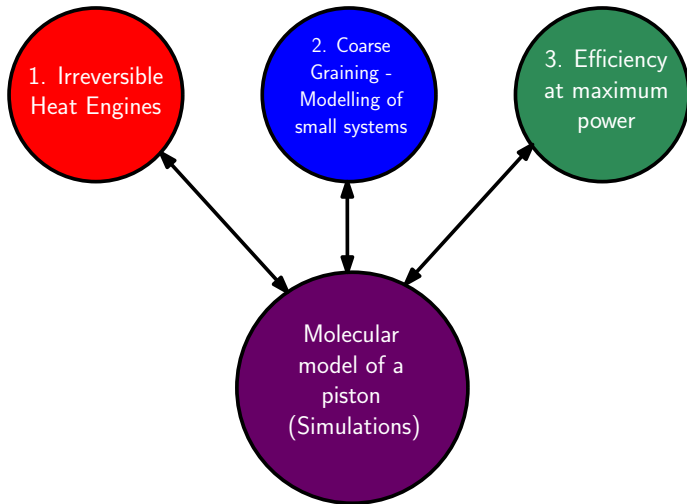
in collaboration with L. Cerino and A. Puglisi

Dip Fisica - Univ. Sapienza Roma and
Centro Interdisciplinare "B Segre", Accademia dei Lincei

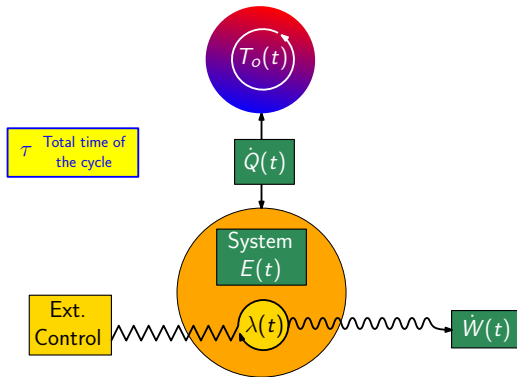
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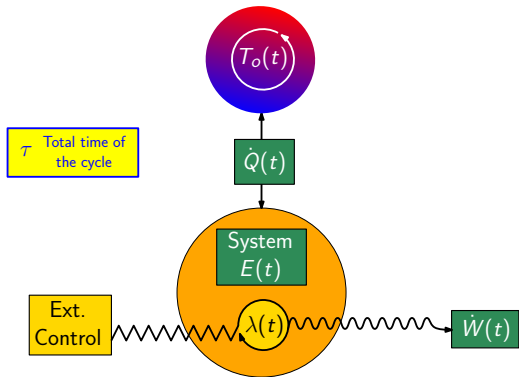
Outline of the talk



Heat Engines: general considerations



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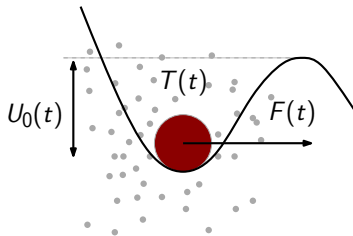
Desiderata

- ▶ Predict the dependence of the integrated fluxes W and Q on τ ;
- ▶ Take into account fluctuations (e.g. predict $P(W)$);

Beyond standard thermodynamics: two possible approaches

Stochastic Thermodynamics

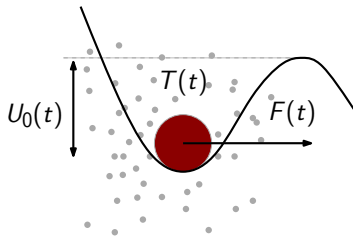
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- ▶ External time-dependent potential
- ▶ Interaction with a reservoir (thermal noise)



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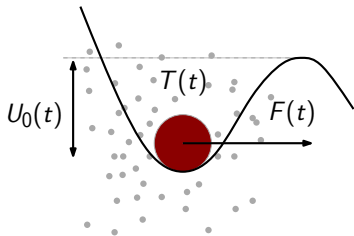
Macroscopic Fluct. Theory

- ▶ $N \gg 1$ particles
- ▶ Hydrodynamical description:
 - ▶ density field $\rho(\mathbf{x}, t)$,
 - ▶ velocity field $u(\mathbf{x}, t)$,
 - ▶ current field $j(\mathbf{x}, t)$.
- ▶ Thermodynamics \Leftrightarrow external fields and special boundary conditions (*thermostats*).

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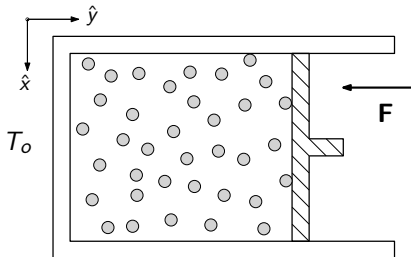
...in the middle?

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A paradigmatic small system

A system composed of $N \sim \mathcal{O}(10^2)$ degrees of freedom



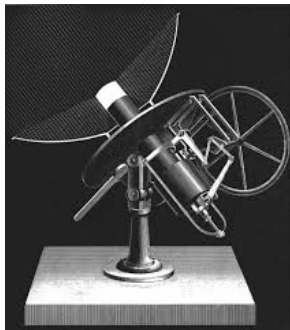
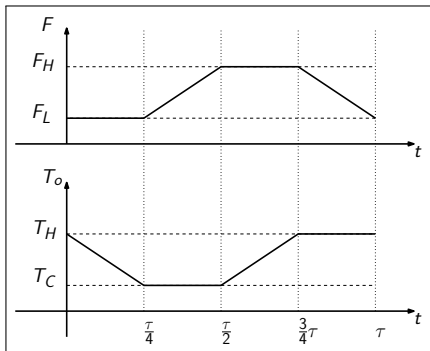
$$\mathcal{H} = \sum_{i=1}^N \frac{p_i^2}{2m} + \frac{P^2}{2M} + FY$$

(+ elastic collision
between particles and
piston)

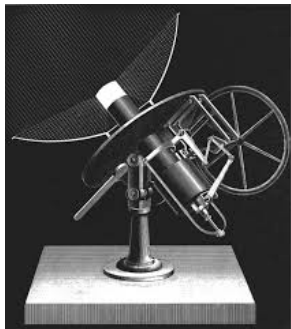
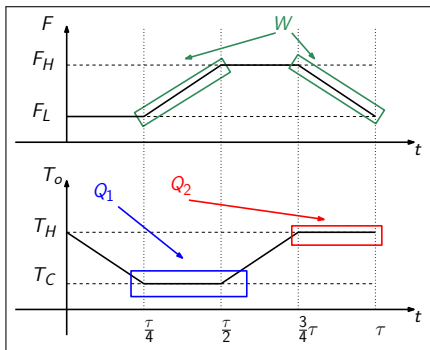
(+ thermal wall on the left
side at temperature T_o)

Is it possible to extract mechanical work from this system with a cyclical protocol?

Heat Engine: the Ericsson cycle



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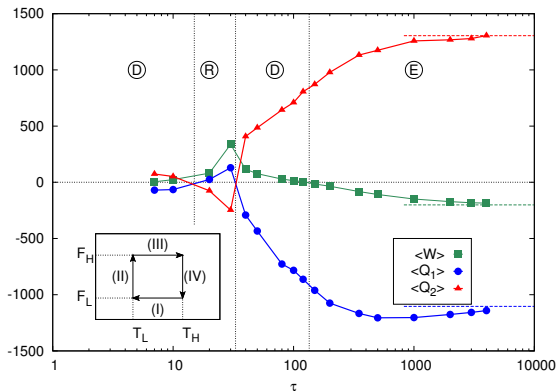
In each segment:

$$W = \int dt \frac{\partial \mathcal{H}}{\partial t} = \int dt \dot{F} X(t)$$

$$Q = \Delta \mathcal{H} - W$$

Results of MD simulations

[L.Cerino, A. Puglisi and A. Vulpiani, PRE E **91**, 032128 (2015)]



Thermodynamics
forces:

$$\delta = \frac{T_H - T_C}{T_H + T_C} = 0.08$$

$$\epsilon = \frac{F_H - F_L}{F_H + F_L} = 0.1$$

Coarse-graining: can we understand this
behavior?

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Identify the relevant (slow-varying) variables of the system.

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Derive a set of **coupled** Langevin equations for these variables;

► Step 3

Use stochastic thermodynamics to derive an explicit expression for thermodyn. quantities ($W, Q, \eta \dots$) and associated fluctuations.

Model with 3 Macroscopic Variables

A coarse grained description is possible in terms of:

- ▶ X piston position
- ▶ V piston velocity
- ▶ T gas kin. energy per particle

$$\mathbf{y} = (X - X_{eq}(t), V, T - T_{eq}(t))$$

Linear time-dependent Langevin eqn.

$$\dot{\mathbf{y}} = \mathbf{A}(\mathbf{t}) \cdot \mathbf{y} + \mathbf{B}(\mathbf{t}) \cdot \boldsymbol{\xi} \leftarrow \text{white noise}$$

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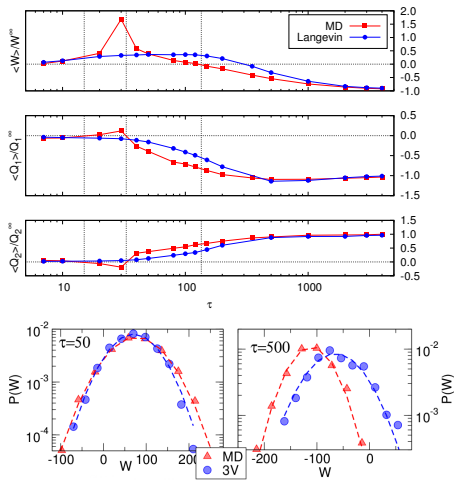
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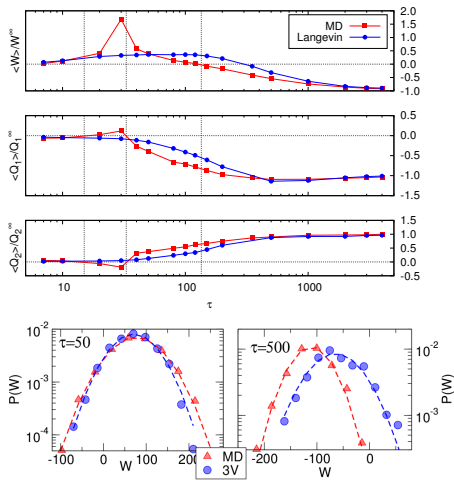
- ▶ \mathbf{A} is determined via kinetic theory considerations (collisions gas \leftrightarrow piston and gas \leftrightarrow thermostat).
- ▶ \mathbf{B} is determined a *fortiori* to restore detailed balance with equilibrium distribution.

Comparison with MD simulations



- ▶ Av. values: Good qualitative agreement (sign inversion, maximum...)
- ▶ Fluctuations: Approx. Gaussian behaviour and same stdev.

Comparison with MD simulations



- ▶ **Av. values:** Good qualitative agreement (sign inversion, maximum...)
- ▶ **Fluctuations:** Approx. Gaussian behaviour and same stdev.
- ▶ **Discrepancies:** Effect of gas inhomogeneities and non-linear effects.

An even simpler model...

We can pass from 3 variables \rightarrow 2 variables by simply fixing $T(t) = T_o(t)$

$$\frac{dX}{dt} = V$$

$$\frac{dV}{dt} = -k(t)(X - X_0(t)) - \gamma(t)V + \sqrt{\frac{2\gamma k_B T_o(t)}{M}} \xi$$

$$k(t) = \frac{F(t)^2(m+M)}{M^2 N k_B T_o(t)} \quad \gamma(t) = \frac{2F(t)}{M} \sqrt{\frac{2m}{\pi k_B T_o(t)}}$$

$$X_0(t) = (N+1) \frac{k_B T_o(t)}{F(t)}$$

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► An important remark

The energy of the system is “imported” from the molecular model: $E(t) = \frac{1}{2}NT(t) + F(t)X(t) + \frac{1}{2}MV^2$.

In principle $E(t)$ is different from the potential
 $U(t) = \frac{1}{2} (MV^2 + k(t)(X - X_0)^2)$.

...to obtain analytic formulas!

With a simpler protocol

$$T_o(t) = T_0 \left[1 + \delta \sin \left(\frac{2\pi t}{\tau} \right) \right]$$
$$F(t) = F_0 \left[1 + \epsilon \cos \left(\frac{2\pi t}{\tau} \right) \right]$$

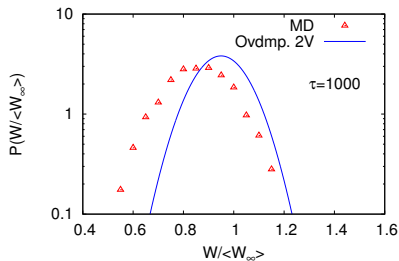
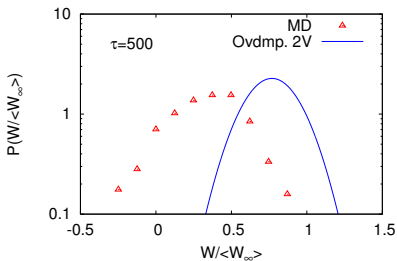
An analytic expression for $P(W)$ (in the **engine regime** for small ϵ and δ): **Gaussian!**

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Linear regime and Onsager-coefficients

- ▶ Total entropy production

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- ▶ Physical interpretation (Brandner, Saito and Seifert arXiv preprint (2015)):

$$J_1 = -\frac{W}{\epsilon T_C}$$
$$J_2 = \frac{T_C + T_H}{T_C T_H} \int_0^\tau \gamma(t) \dot{Q}(t) dt$$

where $\gamma(t)$ is a **smoothing function**. E.g. with two thermostats $\gamma(t) = 1$ if $T = T_H$ and $\gamma(t) = 0$ if $T = T_C$.

Onsager coefficients in the 2V Model

$$\Delta S = J_1 \epsilon + J_2 \delta$$

- ▶ In the linear regime:

$$\begin{pmatrix} J_1 \\ J_2 \end{pmatrix} = \begin{pmatrix} L_{11} & L_{12} \\ L_{21} & L_{22} \end{pmatrix} \begin{pmatrix} \epsilon \\ \delta \end{pmatrix}$$

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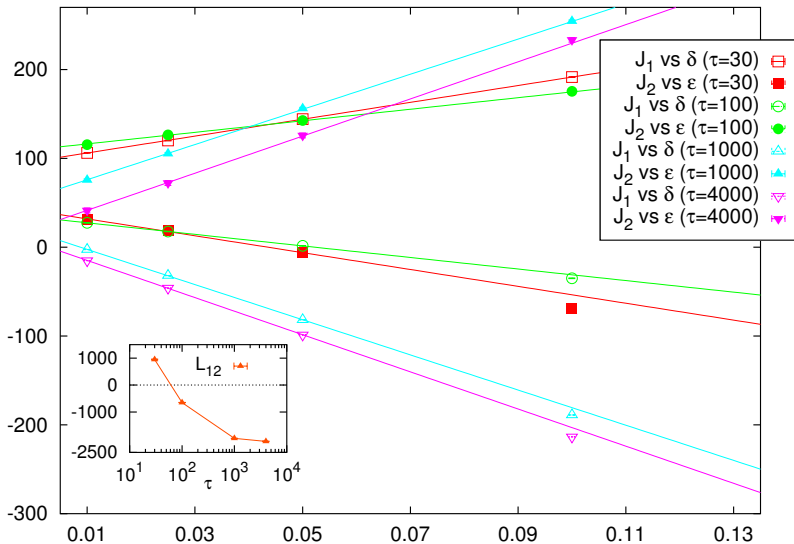
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- ▶ The Onsager coefficients have the right symmetries (derived from [time-reversal](#));
- ▶ $\Delta S \rightarrow 0$ for $\tau \rightarrow \infty$.

Onsager Coeff. in the molecular model



Efficiency at maximum power

- Efficiency:

$$\eta = -\frac{W}{Q_{in}} \leq \eta_C = 1 - \frac{T_C}{T_H} \approx 2\delta$$

where $\delta = \frac{T_H - T_C}{T_H + T_C}$.

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- ▶ $\tilde{\eta}$ = Efficiency at max. power: ...but maximum with respect to which parameter????

Efficiency at maximum power: General Results in the Linear Regime

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- ▶ What about the time τ of the cycle?

Efficiency at τ -maximum power

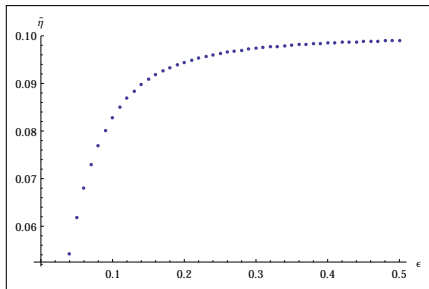
$$\text{Power: } w = -\frac{W}{\tau} = -\epsilon \frac{T_0 J_1}{\tau}$$

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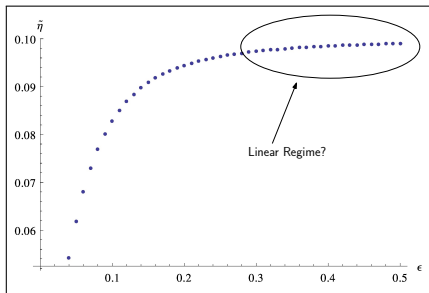


Efficiency at maximum w with respect to τ at different values of ϵ

Efficiency at τ -maximum power

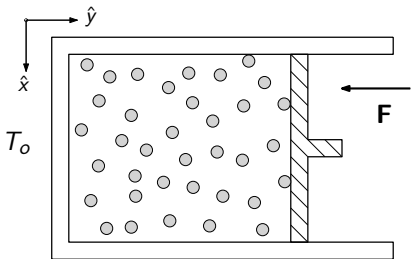
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Efficiency at maximum w with respect to τ at different values of ϵ

Summarizing...



- ▶ Rich phenomenology (due to $N \neq 1$);
- ▶ Fluctuating thermodynamic quantities (due to $N \neq \infty$);
- ▶ Non trivial Langevin description (e.g. impossible to define energy from the Lang. Eq.).

A good insight into the thermodynamics of small systems!

Thank you for the attention!